

A DIELS-ALDER ADDUCT OF N-METHYLFLINDERSINE AND A QUINOLONE
QUINONE METHIDE

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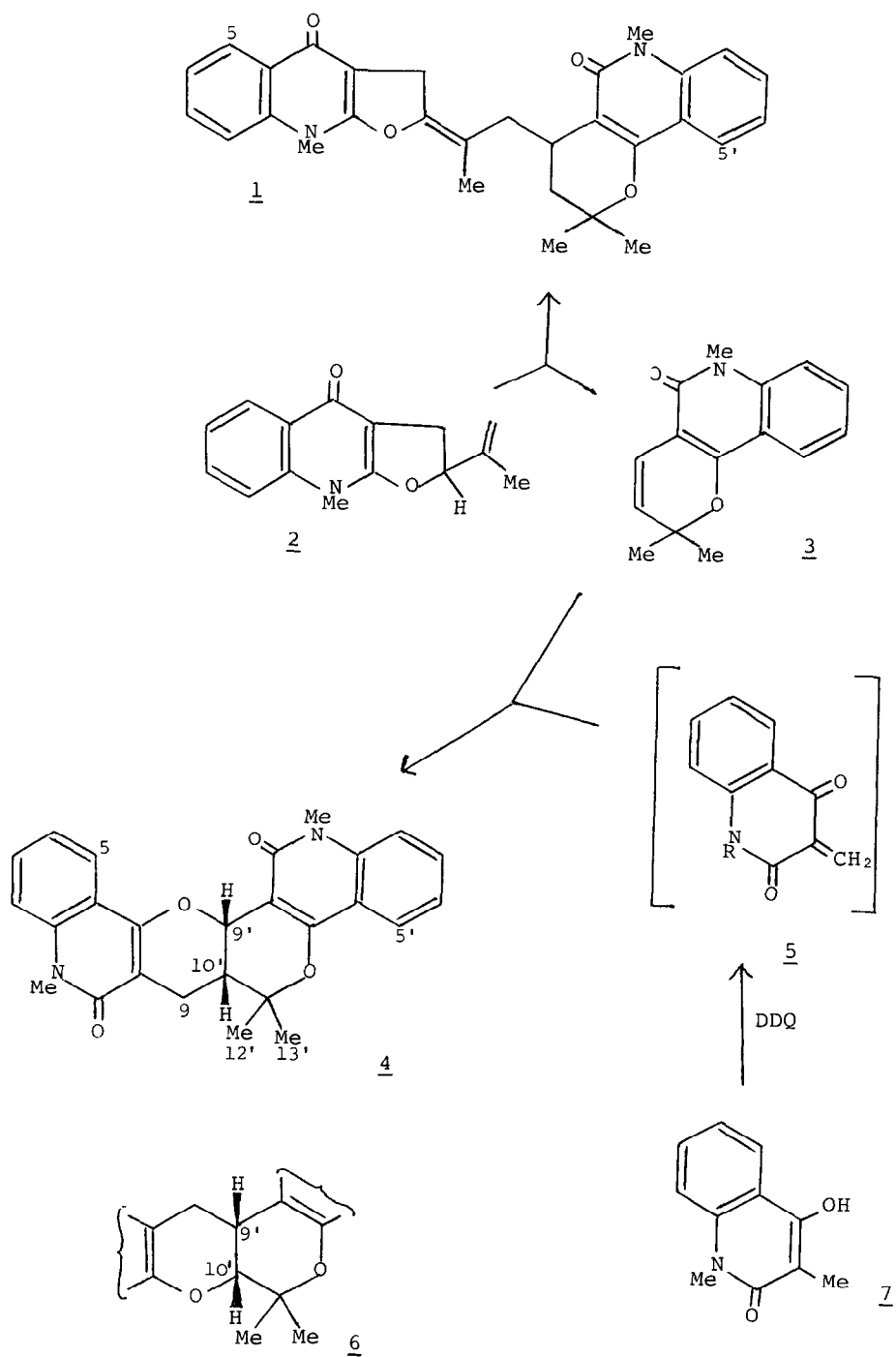
Summary

A quinolone quinone methide (5; R=Me), prepared from 1,3-dimethyl-4-hydroxy-2-quinolone (7) and DDQ, reacted readily with N-methylflindersine (3) to give a single cyclo-addition product (4).

The 'dimeric' quinolone alkaloids of Ptelea trifoliata,¹ for example pteledimerine (1), can be regarded as products of an ene reaction between a dihydrofuroquinolone (2) and N-methylflindersine (3); it is significant that the latter alkaloid and quinolones (2) with oxygen substituents in the aromatic ring were also obtained from P. trifoliata.² A new type of dimeric quinoline alkaloid isolated from Vepris louisii³ may be a Diels-Alder adduct of 7,8-dimethoxy-N-methylflindersine, which was also shown to be a constituent of the plant.⁴ These results, as well as the interesting biological properties of N-methylflindersine,⁵ prompted us to study the parent compound and its derivatives as potential enophiles and dienophiles.

The function of quinolone methides (5) as dienes in cyclo-addition reactions was of particular interest. These species, often in a protonated form, appear as fragment ions when many hemiterpenoid quinolone alkaloids, including pteledimerine (1), are subjected to electron impact.⁶ Furthermore, new syntheses of heteroaromatic compounds developed by Young and his co-workers⁷ take place by decomposition of quinoline derivatives in the presence of aromatic amines apparently via addition of amines to the methides (5). We now report a Diels-Alder reaction between the quinolone methide (5; R=Me) and N-methylflindersine (3).

The required methide (5; R=Me) was generated in situ by refluxing a solution of 1,3-dimethyl-4-hydroxy-2-quinolone (7) and DDQ in benzene in the presence of N-methylflindersine (3).⁸ A single crystalline adduct (4) was



obtained in high yield. The gross structure of the compound was indicated by the mass spectrum which showed a molecular ion peak at m/z 428.171 (68.9%, $C_{26}H_{24}N_2O_4$) and fragment ions at m/z 241 [60.6%, N-methylflindersine (3)] and at m/z 187 [24.2%, methide (5; R=Me)] derived by cleavage of bonds 9-10' and 9'-O. The ^{13}C -NMR spectrum, which was completely assigned, is also consistent with structure (4); off-resonance decoupling confirmed the presence of two Me carbons (C-12' and C-13'), one methylene carbon (C-9) and two sp^3 methine carbons (C-9' and C-10'). Structure (4) was preferred to (6) (which could be formed in an alternative mode of cyclo-addition) on the basis of the 1H -NMR spectrum; thus, a one-proton doublet at δ 5.37 (J 3Hz, H-9') is in accord with the presence of adjacent Ar and ArO groups in compound (4) but should be at higher field in compound (6) (H-10', ArCHCMe₂-). If the 2-quinolone carbonyl group of methide (5; R=Me) had participated in the formation of the adduct a 4-quinolone function would have been present; this possibility is excluded first by the signals in the ^{13}C -NMR spectrum at 163.0 and at 162.3, characteristic of CO groups in 2-quinolones,⁹ and secondly by the presence in the 1H -NMR spectrum of a two-proton doublet at δ 8.02, comparable to the signal in pteledimerine (1) at δ 8.07 (H-5') rather than to the one at 8.32 (H-5) deshielded by the 4-carbonyl group.

Thus, the 2,2-dimethylpyran derivative (3) and the electron-deficient endione (5; R=Me) interact readily and specifically to give a Diels-Alder adduct; further investigations of reactions of this type related to dimeric quinoline alkaloids are in progress.

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